

(Diagnostic Reagent Grade)

T-46

SARCOSINE OXIDASE [SOXG]

from *Bacillus* sp.
(Sarcosine: oxygen oxidoreductase, EC 1.5.3.1)



Preparation and Specification

Appearance : Yellowish amorphous powder, lyophilized
Specific activity : More than 30 U/mg solid

Properties

Substrate specificity	: See Table 1	
Molecular weight	: 41 kDa (gel filtration)	
Isoelectric point	: pH 4.8	
Michaelis constant	: Sarcosine 3.4×10^{-2} M	
Optimum pH	: 7.5–8.5	Figure 1
pH stability	: 8.0–9.5 (50°C, 10 min)	Figure 2
Optimum temperature	: 45–50°C (pH8.0, 20mM Tris-HCl buffer)	Figure 3
Thermal stability	: Stable at 45°C and below (pH 7.5, 10 min)	Figure 4
Effect of various chemicals	: See Table2, Table3	

Applications for Diagnostic Test

This enzyme is useful for enzymatic determination of **creatinine** when coupled with creatinase and creatininase.

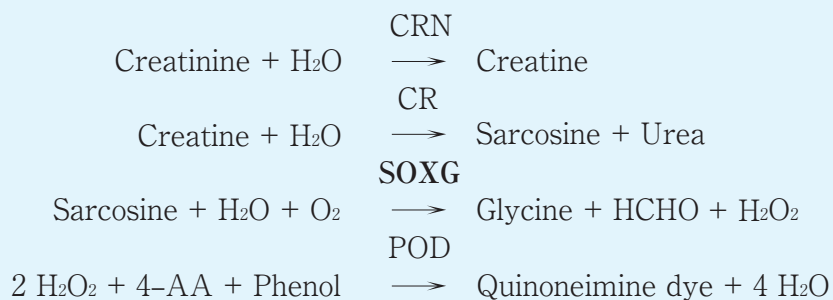


Table 1. Substrate specificity

Substrate	Relative activity (%)
Sarcosine	100
N-ethylglycine	11
Formylglycine	0
N,N-dimethylglycine	0
Glycine	0
Proline	0

Table 2. Effect of various chemicals on SOXG stability (55°C, 10 min)

Additive	Concentration	Residual activity (%)
None		41
FAD	20 μ M	30
KCl	0.3M	101
FMN	10 μ M	21
EDTA	1mM	20
Sucrose	20%	80
Ethylene glycol	20%	2
Glycerol	20%	61

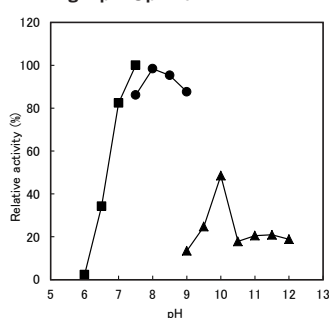
FAD: Flavin adenine dinucleotide

FMN: Flavin mononucleotide

Table 3. Effect of various chemicals on SOXG activity

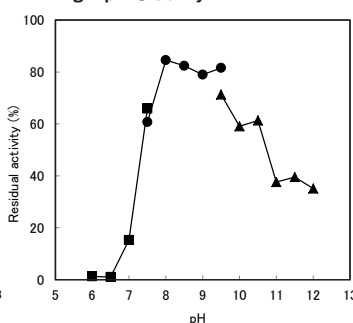
Additive	Concentration	Relative activity (%)
None		100
MgCl ₂	0.5mM	99
MnCl ₂	0.5mM	102
CaCl ₂	0.5mM	97
LiCl ₂	0.5mM	96
CuCl ₂	0.5mM	94
Ba(CH ₃ COO) ₂	0.5mM	100
NaCl	0.5mM	98
CoCl ₂	0.5mM	76
FeCl ₂	0.5mM	76
KCl	0.5mM	96
EDTA	1.0mM	98
Triton X-100	0.1%	99
Sodium Cholate	0.1%	92
Tween 80	0.1%	102

Fig.1 pH Optimum



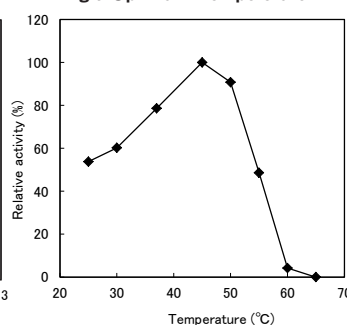
■ : PIPES buffer
 ● : Tris-HCl buffer
 ▲ : Glycine-NaOH buffer

Fig.2 pH Stability



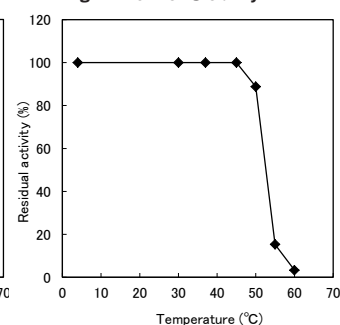
■ : PIPES buffer
 ● : Tris-HCl buffer
 ▲ : Glycine-NaOH buffer

Fig.3 Optimum Temperature



pH 8.0
 20mM Tris-HCl buffer

Fig.4 Thermal Stability

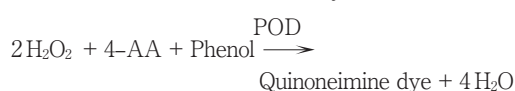


pH 7.5, 10min
 10mM Phosphate buffer

Assay

Principle

The assay is based on the increase in absorbance at 480 nm as the formation of quinoneimine dye proceeds in the following reactions:



Unit definition

One unit is defined as the amount of enzyme which oxidizes 1 μ mole of sarcosine to glycine per minute at 37°C under the conditions specified in the assay procedure.

Reagents

- Reaction mixture

0.2M Tris-HCl buffer pH 8.0	0.05 ml
1.0M Substrate solution (Sarcosine)	0.10 ml
100U/ml POD solution ¹⁾	0.025 ml
15mM 4-AA solution	0.05 ml
0.2% (W/V) Phenol solution	0.05 ml
Distilled water	0.225 ml

1): 100 U/ml POD solution

Dissolve 1,000 U (PPU) of POD with 10 ml of distilled water.

- Reaction stopper

Ethanol	
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- Enzyme dilution buffer

10 mM KH ₂ PO ₄ -K ₂ HPO ₄ buffer pH 7.5	
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- Reagents

Sarcosine (N-methylglycine or methylaminoacetate):

Tokyo Kasei Kogyo Co., Ltd. Special grade #M0332

4-AA: NACALAI TESQUE, INC. Special grade #01907-52
POD: Sigma Chemical Co. Type II # P-8250

■ Enzyme solution

Accurately weigh about 20 mg of the sample and add enzyme dilution buffer to make a total of 20 ml. Dilute it with enzyme solution buffer to adjust the concentration as required.

■ Procedure

1. Pipette accurately 0.5 ml of reaction mixture into a small test tube and preincubate at 37°C.
2. After 5 min, add exactly 10 μ l of enzyme solution and mix to start the reaction at 37°C.
※ In the case of a test blank, add 10 μ l of enzyme dilution buffer in place of enzyme solution.
3. At 5 min after starting the reaction, add 2.50 ml of the reaction stopper to stop the reaction.
4. Measure the absorbance at 480 nm.

$$\begin{aligned} \text{Absorbance sample} &: A_s \\ \text{blank} &: A_b \\ \Delta A &= (A_s - A_b) \leq 0.125 \text{ Abs} \end{aligned}$$

■ Calculation

$$\text{Activity (U/mg of powder)} = \frac{\Delta A / 5}{17.14 \times 1/2} \times \frac{3.01}{0.01} \times \frac{1}{X}$$

17.14 : millimolar extinction coefficient of quinoneimine dye at 480 nm ($\text{cm}^2 / \mu\text{mole}$)

1/2 : a multiplier derived from the fact that 2 mole of H_2O_2 produces 1 mole of quinoneimine dye

5 : reaction time (min)

3.01 : final volume (ml)

0.01 : volume of enzyme solution (ml)

X : concentration of the sample in enzyme solution (mg/ml)

Storage

Storage at -20°C in the presence of a desiccant is recommended. The enzyme activity will be retained for at least one year under this condition.

References

1. Mori, N., Sano, M., Tani, Y. and Yamada, H. (1980) Agric. Biol. Chem., **44**, 1391-1397.
2. Suzuki, M. and Yoshida, M. (1976) Proceedings of the Symposium on Chemical Physiology and Pathology (Kyoto), Vol. **16**, 220.
3. Suzuki, M. (1981) J. Biochem., **89**, 599-607.
4. Kinoshita, T. and Hiraga, Y. (1980) Chem. Pharm. Bull., **28**, 3501-3506.

SOXG 活性測定法 (Japanese)

I. 試薬液

1. 反応試薬混合液
0.2M トリス-HCl 緩衝液 pH8.0 0.05 ml
1M 基質溶液 (サルコシン) 0.10 ml
15mM 4-AA 溶液 0.05 ml
0.2% (W/V) フェノール液 0.05 ml
100U/ml POD 溶液¹⁾ 0.025 ml
精製水 0.225 ml

1): 100U/ml POD 溶液

POD 1,000 単位 (PPU) を精製水 10ml で溶解する。

2. 反応停止液
エタノール原液を用いる。

3. 酵素溶解希釈用液
10mM KH_2PO_4 - K_2HPO_4 緩衝液 pH7.5

4. 試薬
サルコシン (N-メチルグリシン又はメチルアミノ酢酸): 東京化成製 特級 #M0332
4-AA: ナカライテスク製 特級 #01907-52
POD: シグマ製 Type II #P-8250

II. 酵素試料液

検品約 20mg を精密に量り、酵素溶解希釈用液で全容 20ml とする。

その液を酵素溶解希釈用液で適宜希釈する。

III. 測定操作法

1. 小試験管反応試薬混合液 0.50ml を正確に分注し、 37°C で予備加温する。
2. 5 分経過後、酵素試料液 10 μ l を正確に加えて混和し、 37°C で反応を開始する。
※盲検は酵素試料液の代わりに酵素溶解希釈用液 10 μ l を加える。
3. 5 分経過後、反応停止液 2.50ml を加えて混和し、反応を停止する。
4. 480nm における吸光度を測定する。
求められた吸光度の試料液は A_s 、盲検液は A_b とする。

$$\Delta A = (A_s - A_b) \leq 0.125 \text{ Abs}$$

IV. 計算

$$\text{活性 (U/mg)} = \frac{\Delta A / 5}{17.14 \times 1/2} \times \frac{3.01}{0.01} \times \frac{1}{X}$$

17.14: キノンイミン色素の 480nm におけるミリモル分子吸光係数 ($\text{cm}^2 / \mu\text{mole}$)

1/2 : H_2O_2 モルからキノンイミン色素 1 モルが生成することによる係数

5 : 反応時間 (min)

3.01 : 反応総液量 (ml)

0.01 : 反応に供した酵素試料液量 (ml)

X : 酵素試料液の検品濃度 (mg/ml)