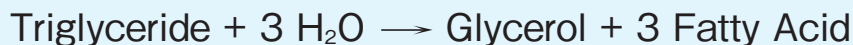


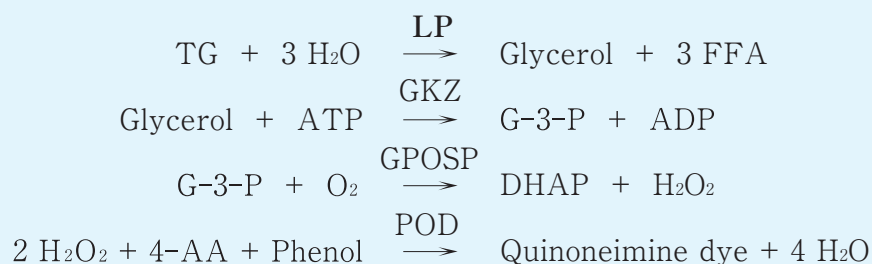
from *Chromobacterium viscosum*  
(Triacylglycerol acylhydrolase, EC 3.1.1.3)  
(Triacylglycerol lipase)



|                     |                                                    |
|---------------------|----------------------------------------------------|
| Appearance          | : White to off-white amorphous powder, lyophilized |
| Specific activity   | : More than 2,500 U/mg solid                       |
| Contaminants        | :                                                  |
| Cholesterol oxidase | Less than 0.01 % (U/U)                             |
| Catalase            | Less than 0.01 % (U/U)                             |

|                             |                                            |          |
|-----------------------------|--------------------------------------------|----------|
| Substrate specificity       | : See Table 1                              |          |
| Molecular weight            | : 120 kDa (pH 3.7) (gel filtration)        |          |
|                             | 30 kDa (pH 7.3) (gel filtration)           |          |
| Isoelectric point           | : pH 3.7 and pH 7.3                        |          |
| Optimum pH                  | : 3.0–10.0                                 | Figure 1 |
| pH stability                | : 4.0–10.0 (50°C, 60 min)                  | Figure 2 |
| Thermal stability           | : Stable at 70°C and below (pH 7.0, 10 hr) | Figure 3 |
| Storage stability           | : At least one year at –20°C               | Figure 4 |
| Effect of various chemicals | : See Table 2                              |          |
| Activators                  | : Detergents                               |          |

This enzyme is useful for enzymatic determination of **triglyceride** when coupled with glycerophosphate oxidase (T-60) and glycerol kinase (T-64).



DHAP: Dihydroxyacetone phosphate

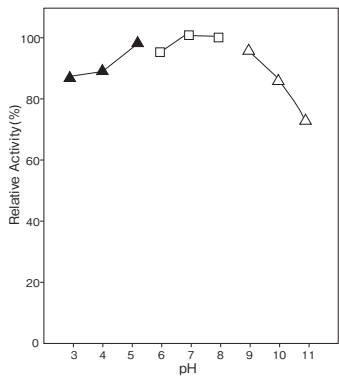
Table 1. Substrate specificity

| Substrate    | Relative activity (%) |
|--------------|-----------------------|
| Triolein     | 100                   |
| Tripalmitin  | 22                    |
| Trimyristin  | 53                    |
| Trilaurin    | 103                   |
| Tricaprin    | 166                   |
| Tricaprylin  | 312                   |
| Tricaproin   | 156                   |
| Tributylin   | 94                    |
| Tripropionin | 22                    |
| Triacetin    | 38                    |

Table 2. Effect of various chemicals on LP activity

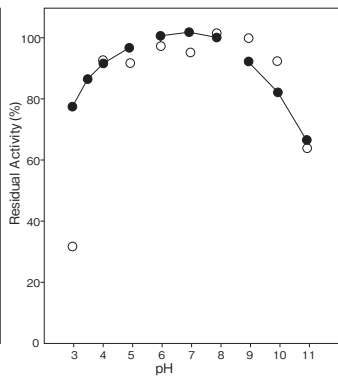
| Additives                                       | Concentration | Relative activity (%) |
|-------------------------------------------------|---------------|-----------------------|
| None                                            | -             | 100                   |
| NiCl <sub>2</sub>                               | 1mM           | 97                    |
| MnCl <sub>2</sub>                               | 1mM           | 98                    |
| (NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub> | 1mM           | 100                   |
| MgCl <sub>2</sub>                               | 1mM           | 99                    |
| ZnCl <sub>2</sub>                               | 1mM           | 94                    |
| ZnSO <sub>4</sub>                               | 1mM           | 94                    |
| Ba(CH <sub>3</sub> COO) <sub>2</sub>            | 1mM           | 98                    |
| CaCl <sub>2</sub>                               | 1mM           | 100                   |
| MoSO <sub>4</sub>                               | 1mM           | 106                   |
| CuSO <sub>4</sub>                               | 0.5mM         | 26                    |
| CuCl <sub>2</sub>                               | 0.5mM         | 26                    |
| FeCl <sub>3</sub>                               | 1mM           | 103                   |
| CoCl <sub>2</sub>                               | 1mM           | 96                    |
| Li <sub>2</sub> CO <sub>3</sub>                 | 1mM           | 93                    |
| EDTA                                            | 1mM           | 105                   |
| KCl                                             | 100mM         | 100                   |
| NaCl                                            | 100mM         | 98                    |
| NaN <sub>3</sub>                                | 0.05%         | 97                    |
| NaF                                             | 20mM          | 88                    |

Fig.1 pH Optimum



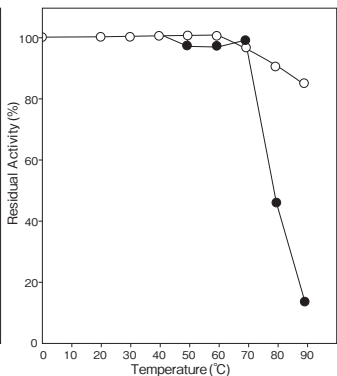
▲ : McIlvaine buffer  
□ : Phosphate buffer  
△ : Borate buffer

Fig.2 pH Stability



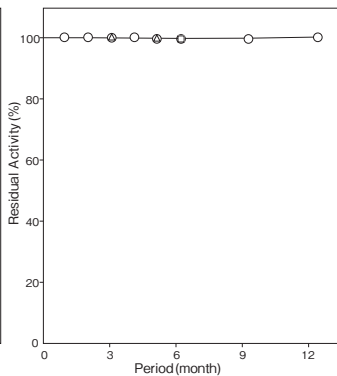
● : 37°C, 24 hr  
○ : 50°C, 1 hr  
pH 3, 4, 5 McIlvaine buffer  
pH 6, 7, 8 Phosphate buffer  
pH 9, 10, 11 Borate buffer

Fig.3 Thermal Stability



pH 7.0, 10 hr  
○ : Lyophilized powder  
● : Aqueous solution

Fig.4 Storage (lyophilized powder)

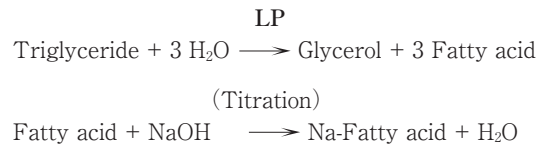


○ : -20°C  
□ : 5°C  
△ : 30°C

## Assay

### Principle

The assay is based on the titration of fatty acids liberated in the following reactions:



### Unit definition

One unit is defined as the amount of enzyme which liberates 1  $\mu$  mole of fatty acid per minute at 37°C under the conditions specified in the assay procedure.

### Reagents

1. Substrate suspension (Olive oil and Adekatol SO-120)  
50 g of Olive oil (Japanese Pharmacopoeia grade) and 50 g of Adekatol SO-120 are suspended with 150 ml of distilled water.

### 2. Reaction stopper

Mixture of ethanol and acetone (1:1)

### 3. Indicator

1% (W/V) Phenolphthalein-ethanol solution

### 4. Titration solution

50 mM NaOH solution

### 5. Enzyme dilution buffer

0.1 M KH<sub>2</sub>PO<sub>4</sub>-NaOH buffer, pH 8.0 containing 0.1% (W/V) BSA and 0.1% (W/V) NaN<sub>3</sub>

### 6. Reagents

Olive oil: (Japanese Pharmacopoeia grade)  
Ethanol: (Japanese Pharmacopoeia grade)  
Adekatol SO-120 : ADEKA CORPORATION  
BSA: Millipore Fraction V pH5.2 #81-053

### Enzyme solution

Accurately weigh about 10 mg of the sample and add enzyme dilution buffer to make a total of 50 ml. Dilute it with enzyme dilution buffer to adjust the concentration to within 2-4 U/ml.

## ■ Procedure

1. Pipette accurately 5 ml of substrate suspension and 2 ml of distilled water into a test tube (24 mm i.d. × 200 mmH) and mix to start the preincubation at 37°C.
2. After 10 min, add 0.5 ml of enzyme solution and mix to start the reaction.  
※ In the case of a test blank, add 0.5 ml of enzyme dilution buffer in place of enzyme solution.
3. After 20 min, stop the reaction with 16 ml of reaction stopper.
4. Add 3 drops of indicator and titrate the whole mixture with under nitrogen gas bubbling.  
※ End point of titration: Appearance of the same color as that of the blank

$$\begin{aligned} \text{Titration volume} \quad & \text{sample} : V_s \\ & \text{blank} : V_c \\ \Delta V = (V_s - V_c) & \leq 2.5 \text{ ml} \\ V_c & \leq 0.6 \text{ ml} \end{aligned}$$

## ■ Calculation

$$\text{Activity (U/mg of powder)} = \frac{\Delta V \times F}{20} \times 50 \times \frac{1}{0.5} \times \frac{1}{X}$$

20 : reaction time (min)

F : factor of titration solution (50 mM NaOH)

50 : concentration (mM) of titration solution (50 mM NaOH)

0.5 : the volume of enzyme solution (ml)

X : concentration of the sample in enzyme solution (mg/ml)

## Storage

Storage at -20°C in the presence of a desiccant is recommended. Enzyme activity will be retained for at least one year under this condition (Figure 4).

## References

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5. Horiuchi, Y., Koga, H. and Gocho, S. (1976) J. Biochem. (Tokyo), **80**, 367-370.
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## LP 活性測定法 (Japanese)

### I. 試薬液

1. 基質懸濁液 (オリーブ油とアデカトル SO-120 の懸濁液)  
「局方」オリーブ油 50.0g とアデカトル SO-120 50.0g を精製水 150ml に懸濁する。
2. 反応停止液  
エタノール-アセトン (1:1) 混液
3. 指示液  
1% (W/V) フェノールフタレン-エタノール溶液
4. 滴定液  
50mM NaOH 液
5. 酵素溶解希釈用液  
0.1% (W/V) BSA と 0.1% (W/V)  $\text{NaN}_3$  を含む 0.1M  $\text{KH}_2\text{PO}_4$ -NaOH 緩衝液 pH8.0
6. 試薬  
オリーブ油:「局方」  
エタノール:「局方」  
アデカトル SO-120: ADEKA 製  
BSA: Millipore 製 Fraction V pH5.2 #81-053

### II. 酵素試料液

検品約 10mg を精密に量り、酵素溶解希釈用液に溶解して全容 50ml とする。  
その液を酵素溶解希釈用液で 2~4U/ml 濃度となるように適宜希釈する。

### III. 測定操作法

1. 試験管 (24mm i.d. × 200mm H) に基質懸濁液 5ml と精製水 2ml を正確に分注して攪拌混和後、37°C で予備加温する。
2. 10 分経過後、酵素試料液 0.5ml を加えて混和し、37°C で反応を開始する。  
※ 盲検は酵素試料液の代わりに酵素溶解希釈用液 0.5ml を加える。
3. 20 分経過後、反応停止液 16ml を加えて反応を停止する。
4. 指示液 3 滴を加えて  $\text{N}_2$  ガスで攪拌しながら滴定液で滴定する。  
※ 滴定の終点は盲検時と同色を呈した時点とする。  
求められた滴定量を試料液は  $V_s$ 、盲検液は  $V_c$  とする。

$$\begin{aligned} \Delta V = (V_s - V_c) & \leq 2.5 \text{ ml} \\ V_c & \leq 0.6 \text{ ml} \end{aligned}$$

### IV. 計算

$$\text{活性 (U/mg)} = \frac{\Delta V \times F}{20} \times 50 \times \frac{1}{0.5} \times \frac{1}{X}$$

20 : 反応時間 (min)

F : 滴定液 (50mM NaOH) の Factor

50 : 滴定液 (50mM NaOH) の濃度 (mM)

0.5 : 反応に供した酵素試料液量 (ml)

X : 酵素試料液中の検品濃度 (mg/ml)